



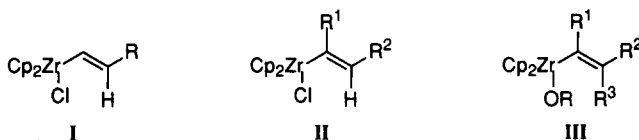
Coupling Reaction of Alkenylzirconocenes with Aryl or Alkenyl Iodides in the Presence of CuCl/Pd(PPh₃)₄

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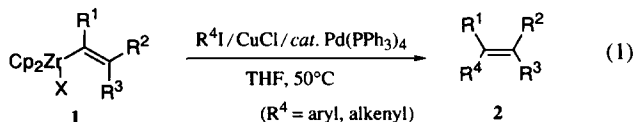
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Abstract: Trisubstituted alkenylalkoxyzirconocenes which have been prepared from carbozirconation reactions of internal alkynes via zirconacyclopentenes reacted with aryl or alkenyl iodides in the presence of CuCl and a catalytic amount of Pd(PPh₃)₄ to give coupling products.
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Alkenylzirconocenes have been useful intermediates for preparation of disubstituted or trisubstituted stereodefined alkenes,^{1,2} since alkenylzirconocenes of type I and II are easily obtained by hydrozirconation reactions of alkynes,³ or by oxidative addition of alkenyl halides to zirconocene.⁴ Further coupling reactions of these alkenylzirconocenes with aryl iodides or alkenyl iodides have been developed by Negishi using Pd(PPh₃)₄ (for type I)^{2a} and ZnCl₂/Pd(PPh₃)₄ (for type II)⁵ as the catalyst systems. Recently, we have reported a novel preparative method of alkenylalkoxyzirconocenes (type III: **1a-1c**) and alkenylbromozirconocenes (**1d**) by various carbozirconation reactions of internal alkynes via zirconacyclopentenes. The reactions of zirconacyclopentenes with alcohols, allyl ethers, vinyl ethers and 4-bromo-1-butene gave ethylzirconation (**1a**),⁶ allylzirconation (**1b**),⁷ vinylzirconation (**1c**)⁸ and cyclopropylmethylzirconation (**1d**)⁹ products of alkynes, respectively. However, for type III, unfortunately neither Pd(PPh₃)₄ nor ZnCl₂/Pd(PPh₃)₄ gave positive results.

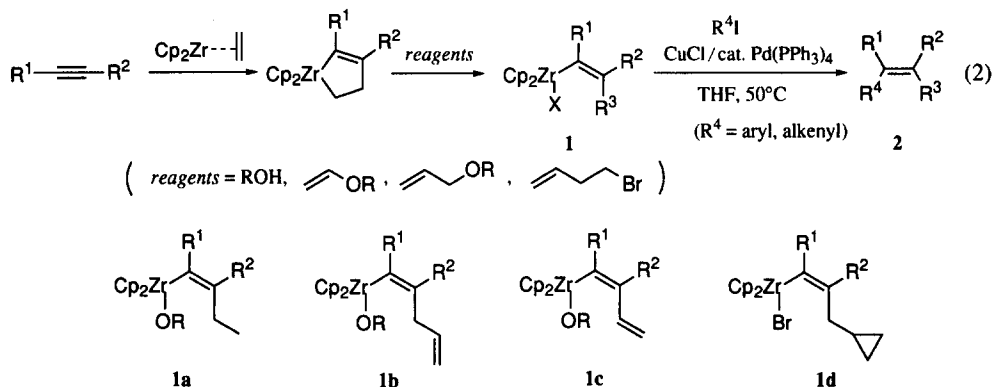


During the course of our study on the reactivity of type III,¹⁰ we found that CuCl/Pd(PPh₃)₄ was a very useful combination for the coupling reaction of type III with aryl or alkenyl iodides to give stereodefined tetrasubstituted alkenes (eq 1).¹¹ In this paper we would like to report the coupling reaction of alkenylalkoxyzirconocenes of type III and alkenylbromozirconocenes with aryl or alkenyl iodides in the presence of CuCl and a catalytic amount of Pd(PPh₃)₄.



Typical procedure is as follows. A solution of zirconacyclopentene¹² prepared from 4-octyne (1

mmol) was treated with (a) *t*-butanol (1.1 mmol) at room temperature for 1h, (b) allyloxytrimethylsilane (2 mmol), at 50°C for 1h, (c) butyl vinyl ether (2 mmol) at 50°C for 6h, or (d) 4-bromo-1-butene (1.5 mmol) at 50°C for 6h gave the corresponding carbozirconation products (**1a-d**). To the solution were added CuCl (1.3 mmol), iodobenzene (1 mmol) and Pd(PPh₃)₄ (0.05 mmol) in this order, and the mixture was stirred at 50°C for 3h. After the usual workup, compounds **2a**¹³ (90%), **2k** (92%), **2o** (67%) and **2q** (77%) were obtained (eq 2). All were one-pot reactions.



The results are summarized in Table 1. To an alkyne were introduced two newly formed carbon-carbon bonds, one by the carbozirconation reaction and the other by a coupling reaction with aryl or alkenyl iodides. The final products, tetrasubstituted alkenes **2a-r**, were obtained in good to high yields. The carbozirconation reaction proceeded selectively in a *syn*-addition manner as reported⁷, and the stereochemistry was retained during the coupling reaction. When allyloxytrimethylsilane was used, allylated compounds **2a-j** were obtained in good to high yields. Since the allylzirconation reaction was not regioselective when 1-phenyl-1-hexyne was employed, a mixture of two regioisomers **2h** was obtained in combined yield of 76% in a ratio of 3:1. In all cases, stereoisomerization of the alkene moiety or homo-coupling¹⁴ of alkenylzirconium **1** was not observed.

Various aryl iodides also successfully coupled with alkenylzirconium compounds of type III. *p*-Substituted phenyl iodides were examined (entry 2-5). The electron-donating or -withdrawing nature of *p*-substituents did not have significant influence on this reaction. 2-Iodothiophene and α -iodonaphthalene gave thienyl compound **2m** and naphthalene derivative **2n**, respectively. Instead of aryl iodides, alkenyl iodides were also employed to give pentasubstituted conjugated dienes **2i**, **2j** and **2l**.

In addition to type III compounds, this catalyst system of CuCl/Pd(PPh₃)₄ could be used for dienylyzirconium compound **4**. Zirconacyclopentadiene **3** was monoprotonated by alcoholysis to give dienylyzirconium complex **4**, and further treatment with iodobenzene under similar reaction conditions gave the coupling product **5** in 84% yield (eq 3).

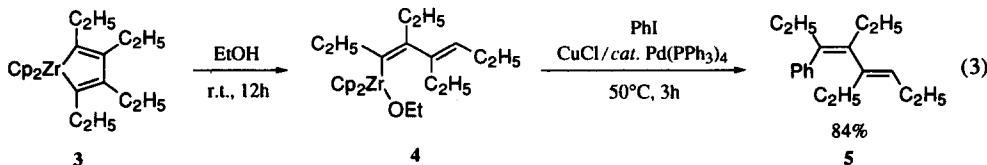


Table 1. Coupling Products of Alkenylzirconocenes with Aryl or Alkenyl Iodides

Entry	R ¹ —≡—R ²	Reagents	R ⁴ I	Products	Yields (%) ^a
1			Y=H		2a 90
2			Y=CH ₃		2b 66
3	C ₃ H ₇ —≡—C ₃ H ₇				2c 74
4			Y=CO ₂ Et		2d 82
5			Y=OMe		2e 74
			Y=NO ₂		
6	Me ₃ Si—≡—C ₄ H ₉		PhI		2f 68
7	Ph—≡—Ph		PhI		2g 51
8	Ph—≡—C ₄ H ₉		PhI		2h 76 (75:25) ^b
9	C ₂ H ₅ —≡—C ₂ H ₅				2i 64
10	C ₃ H ₇ —≡—C ₃ H ₇				2j 81
11	C ₃ H ₇ —≡—C ₃ H ₇	^t BuOH	PhI		2k 92
12	C ₃ H ₇ —≡—C ₃ H ₇	^t BuOH			2l 90
13	C ₃ H ₇ —≡—C ₃ H ₇	^t BuOH			2m 76
14	C ₃ H ₇ —≡—C ₃ H ₇	^t BuOH			2n 76
15	C ₃ H ₇ —≡—C ₃ H ₇		PhI		2o 67
16	Ph—≡—Ph		PhI		2p 66
17	C ₃ H ₇ —≡—C ₃ H ₇		PhI		2q 77
18	Ph—≡—Ph		PhI		2r 84

^a Yields were determined by GC. ^b Ratio of regioisomers.

When the CuCl/Pd(PPh₃)₄ system was applied to alkenylzirconium compounds of type I and II, the desired coupling compounds were the major product. However, a considerable amount of homo-coupling products was observed even in the presence of Pd(PPh₃)₄.

We conclude that, when an alkenylzirconocene has an OR ligand (1a-c, 4), Pd(PPh₃)₄ or ZnCl₂/Pd(PPh₃)₄ was not useful for the coupling reactions of alkenylzirconocenes with aryl or alkenyl iodides. Use of a stoichiometric or catalytic amount of CuCl did not give satisfactory results. However, a combination of CuCl and a catalytic amount of Pd(PPh₃)₄ was very effective.

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11. For the coupling reactions using alkenylcopper, see Regan, A. C. One or More C=C Bond(s) Formed by Addition. In *Comprehensive Organic Functional Group Transformations*, Katritzky, A. R.; Meth-Cohn, O.; Rees, C. W. Eds.; Pergamon Press: Oxford 1995, vol. 1, pp. 539–542.
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13. Compound **2a**: ¹H-NMR (CDCl₃, Me₄Si) δ 0.84 (t, 3H, *J*=7.3Hz), 0.96 (t, 3H, *J*=7.4Hz), 1.19–1.33 (m, 2H), 1.41–1.55 (m, 2H), 2.14 (t, 2H, *J*=7.8Hz), 2.32 (t, 2H, *J*=7.6Hz), 2.57 (d, 2H, *J*=6.3Hz), 4.87–4.94 (m, 2H), 5.61–5.76 (m, 1H), 7.06–7.30 (m, 5H); ¹³C-NMR (CDCl₃, Me₄Si) δ 14.00 (CH₃), 14.32 (CH₃), 21.47 (CH₂), 22.01 (CH₂), 32.88 (CH₂), 36.21 (CH₂), 37.57 (CH₂), 114.91 (CH₂), 125.93 (CH), 127.81 (CH), 128.73 (CH), 133.35 (q), 137.72 (CH), 143.74 (q); HRMS calcd for C₁₇H₂₄, 228.1877, found 228.1870.
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